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Key indicators

Single-crystal X-ray study T = 294 K Mean σ (C–C) = 0.002 Å R factor = 0.046 wR factor = 0.128 Data-to-parameter ratio = 14.8

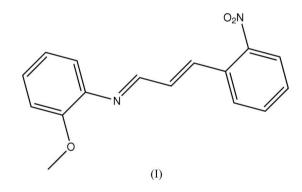
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-Methoxy-N-[3-(2-nitrophenyl)allylidene]aniline

The molecule of the title compound, $C_{16}H_{14}N_2O_3$, is slightly non-planar, with a dihedral angle of 3.3 (1)° between the two benzene rings. In the crystal structure, the molecules are linked into centrosymmetric dimers by intermolecular C– $H \cdots O$ hydrogen bonds.

Comment

We have reported the synthesis and crystal structure of 2-{[3-(2-nitrophenyl)prop-2-enylidene]amino}phenol, (II) (Li *et al.*, 2005). In our ongoing studies of push–pull Schiff bases, the title compound, (I), was obtained. We report here its crystal structure.



The bond lengths and angles are within normal ranges (Allen *et al.*, 1987). All the bonds in (I) show a characteristic length intermediate between those of single and double bonds, and comparable with those in (II). The molecule is slightly non-planar, with a dihedral angle of $3.3 (1)^{\circ}$ between the two benzene rings, in contrast to that of $83.0 (1)^{\circ}$ in (II). There exists an intramolecular C-H···O hydrogen bond (Table 2),

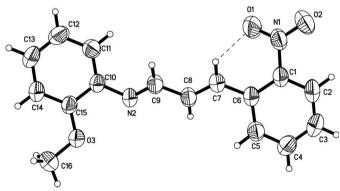


Figure 1

The molecular structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme. The intramolecular hydrogen bond is shown as a dashed line.

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organic papers

forming a six-membered ring (Fig. 1). In the crystal structure, molecules are linked into centrosymmetric dimers by C– $H \cdots O$ intermolecular hydrogen bonds (Table 1 and Fig. 2).

Experimental

Compound (I) was prepared according to the method of Li *et al.* (2005). Single crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of an ethanol–water (2:1) solution over a period of 3 d.

Z = 4

 $D_x = 1.314 \text{ Mg m}^{-3}$ Mo K α radiation $\mu = 0.09 \text{ mm}^{-1}$ T = 294 (2) KBlock, yellow $0.41 \times 0.15 \times 0.11 \text{ mm}$

7553 measured reflections 2807 independent reflections 2042 reflections with $I > 2\sigma(I)$

 $\begin{aligned} R_{\rm int} &= 0.023\\ \theta_{\rm max} &= 26.1^\circ \end{aligned}$

Crystal data

$C_{16}H_{14}N_2O_3$
$M_r = 282.29$
Monoclinic, $P2_1/c$
a = 14.0515 (16) Å
b = 5.0488 (6) Å
c = 20.118 (2) Å
$\beta = 90.706 \ (2)^{\circ}$
V = 1427.1 (3) Å ³

Data collection

Siemens SMART 1000 CCD area-
detector diffractometer
ω scans
Absorption correction: multi-scan
(SADARS Sheldrick 1006)

(SADABS; Sheldrick, 1996) $T_{min} = 0.963, T_{max} = 0.990$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0652P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.046$	+ 0.1418P]
$wR(F^2) = 0.128$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.03	$(\Delta/\sigma)_{\rm max} < 0.001$
2807 reflections	$\Delta \rho_{\rm max} = 0.16 \ {\rm e} \ {\rm \AA}^{-3}$
190 parameters	$\Delta \rho_{\rm min} = -0.16 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$C2-H2A\cdots O2^{i}$	0.93	2.45	3.344 (2)	162
$C7-H7A\cdots O1$	0.93	2.40	2.745 (2)	102

Symmetry code: (i) -x + 1, -y - 1, -z + 1.

All H atoms were located in difference Fourier maps and constrained to ride on their parent atoms, with C-H = 0.96 and 0.93 Å for methyl and other H atoms, respectively, and with $U_{iso}(H) = xU_{eq}(C)$, where x = 1.5 for methyl H atoms and x = 1.2 for other H atoms.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for

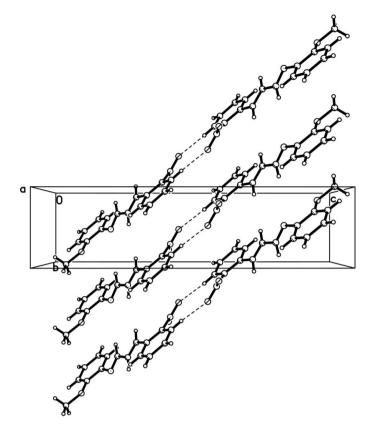


Figure 2

Packing diagram of (I), showing the centrosymmetric dimers. Hydrogen bonds are shown as dashed lines.

publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

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